

Phase Change Materials

Modeling Approach to Facilitate Thermal Energy Management in Buildings

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27.09.2018

COMSOL
CONFERENCE
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Introduction

What are PCMs and what are their application areas?

- Materials with a characteristically large enthalpy of fusion
- Latent heat energy storage systems
- Decouple energy supply and demand → increase efficiency
- Wide range of applications from -40 to 500°C , *i.e.* space to photovoltaics

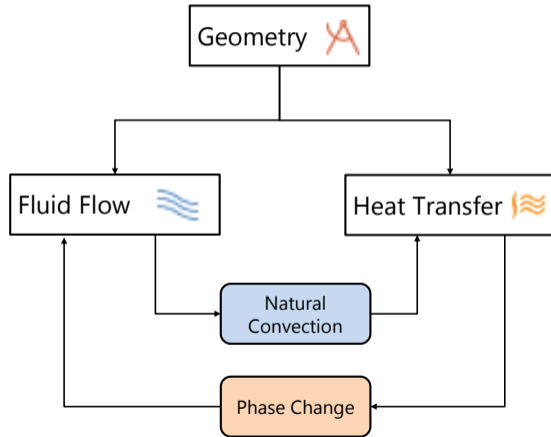
Introduction

The need for modeling PCM

- Obtain fundamental understanding for freezing and melting cycle
- Predict the complex behavior well enough
- Efficiently choose among the vast selection of suitable PCM
- Design improvements
- Reduce development costs

Physical Model

Multiphysical couplings



Numerical Model

Implementation into COMSOL Multiphysics

$$\nabla \cdot \mathbf{u} = 0$$

Laminar Flow (CFD)

$$\rho \frac{\partial \mathbf{u}}{\partial t} + \rho (\mathbf{u} \cdot \nabla) \mathbf{u} = -\nabla p + \nabla \cdot (\mu(T) \nabla \mathbf{u}) + S(T) \mathbf{u} + \mathbf{F}$$

 \mathbf{u} T auxiliary algebraic
equations for material
properties

C_p specific heat capacity at constant pressure
 F volumetric force on fluid
 u velocity
 T temperature
 p pressure
 k thermal conductivity
 t time
 ρ density

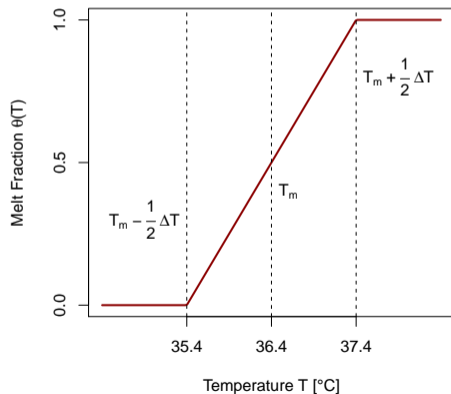
$$\rho(T) C_p(T) \frac{\partial T}{\partial t} + \rho(T) C_p(T) \mathbf{u} \cdot \nabla T = \nabla \cdot (k(T) \nabla T)$$

Heat Transfer in Fluids

$$\theta(T) = \begin{cases} 0, & \text{solid} \\ \frac{T - (T_m - \Delta T/2)}{\Delta T}, & \text{mushy} \\ 1, & \text{liquid} \end{cases} \quad (1)$$

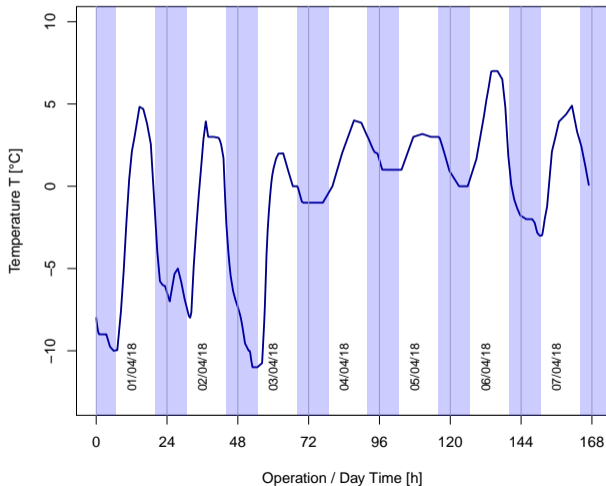
Numerical Model

Melted fraction $\theta(T)$



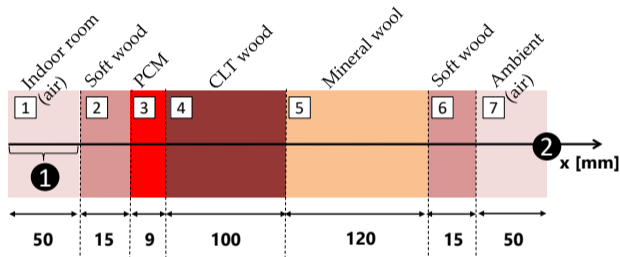
Application Example

Observed week in Oslo, temperature profile



Application Example

Wall crossection of a typical Norwegian building



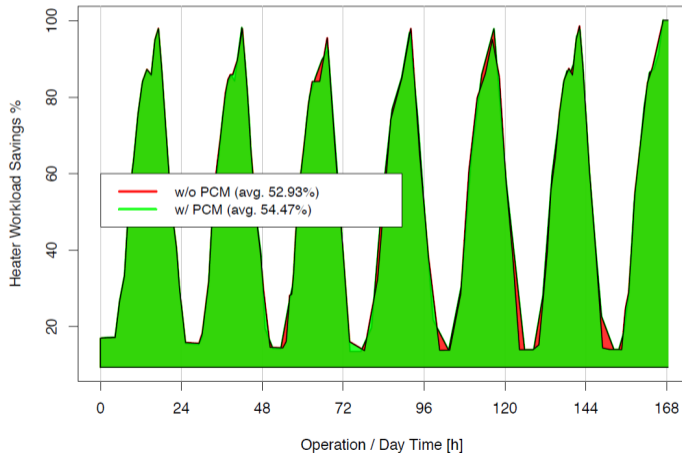
- ① Thermostat: Keep
 $19 < T < 26$ °C
 + internal heat gains

- ② Weather data for T

Layer	Width [mm]	ρ [kgm ⁻³]	C_p [Jkg ⁻¹ K ⁻¹]	k [Wm ⁻¹ K ⁻¹]
1 7	50	1.3	1000	0.13
2 6	15	390	1600	0.13
3 (s)	9	1400	2200	2.5
3 (l)	9	850	4500	0.15
4	100	410	1300	0.098
5	120	60	850	0.04

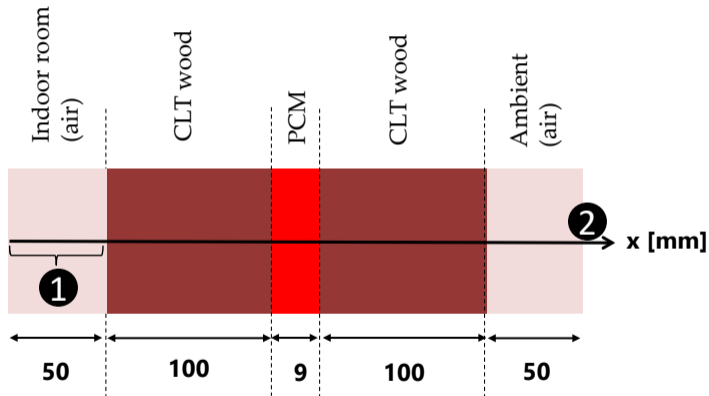
Application Example

Results - energy savings?



Application Example

A less insulated wall crosssection



Introduction

Physical
Model

Numerical
Model

Application
Example

Temperature

Crosssection 1

Results

Crosssection 2

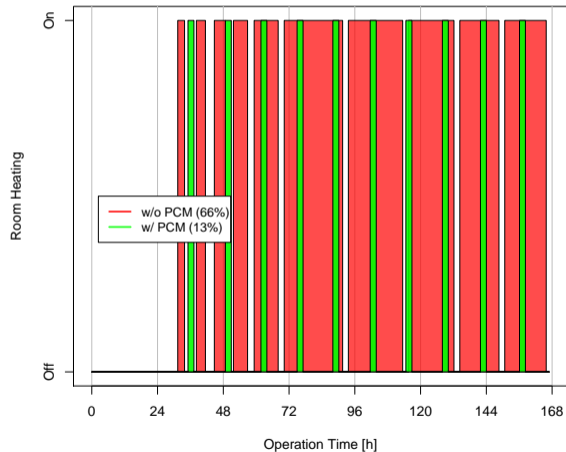
Thermostat

Wall core
temperature

Final Remarks

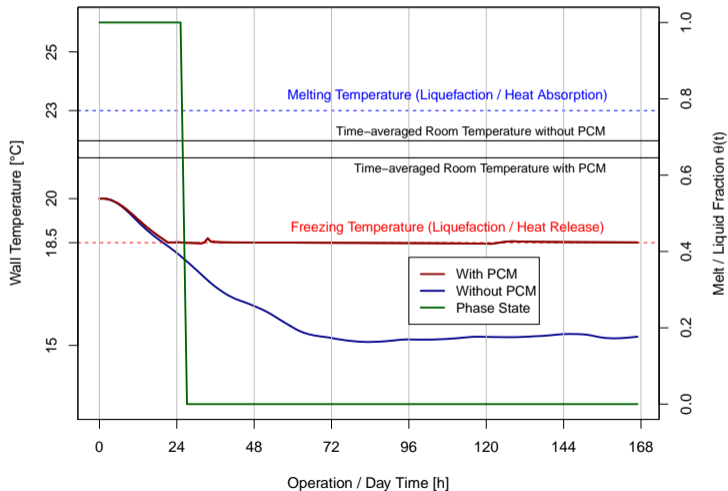
Application Example

Results - substantial energy savings!



Application Example

Wall core temperature



Discussion

- Release heat to reduce heating demand
- For well insulated walls → marginal savings!
- But: PCM reduces peak temperatures on both extremes
- Cold climates → main benefit in summer

Conclusion

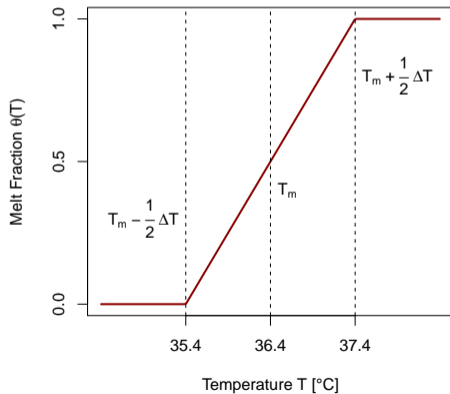
- Comprehensive and suitable modeling approach for phase change phenomena developed
- Rapid orientation whether a PCM meets thermal, technical and economic requirements
- Model shows the importance of including indoor dynamics to assess the PCM potential
- Numerically stable model, extendable to enhanced PCM

Thank you for your attention!

$$\theta(T) = \begin{cases} 0, & \text{solid} \\ \frac{T - (T_m - \Delta T/2)}{\Delta T}, & \text{mushy} \\ 1, & \text{liquid} \end{cases} \quad (2)$$

A - Modeling Functions

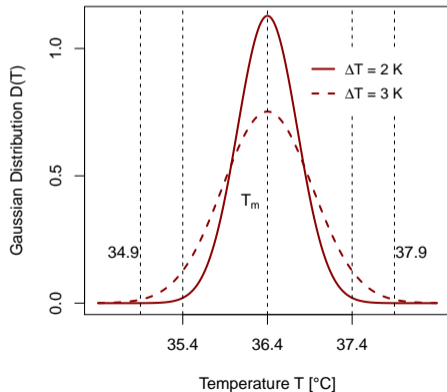
Melt Fraction $\theta(T)$



$$D(T) = \frac{e^{-\frac{(T - T_m)^2}{(\Delta T/4)^2}}}{\sqrt{\pi}(\Delta T/4)^2} \quad (3)$$

A - Modeling Functions

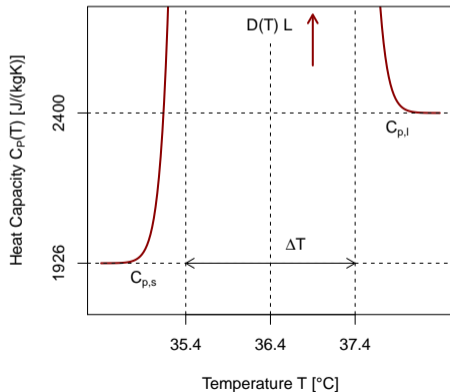
Gaussian Distribution Function $D(T)$



A - Modeling Functions

Modified Heat Capacity $C_p(T)$

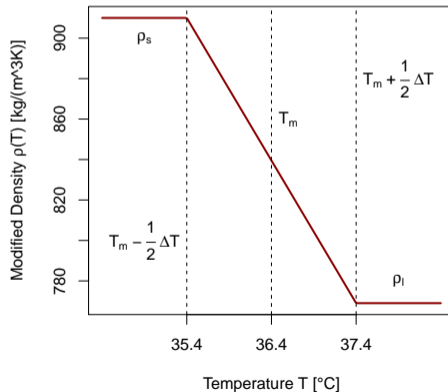
$$C_p(T) = C_{p,s} + \theta(T)(C_{p,l} - C_{p,s}) + D(T)L \quad (4)$$



$$\rho(T) = \rho_s + \theta(T)(\rho_l - \rho_s) \quad (5)$$

A - Modeling Functions

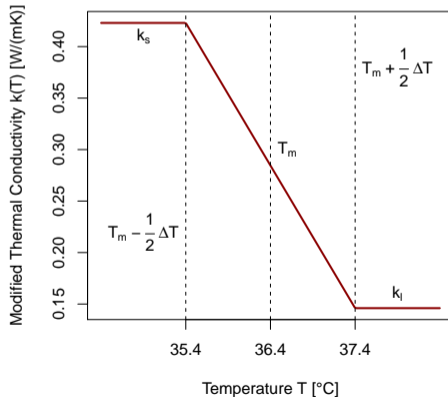
Modified Material Density $\rho(T)$



$$k(T) = k_s + k(T)(k_l - k_s) \quad (6)$$

A - Modeling Functions

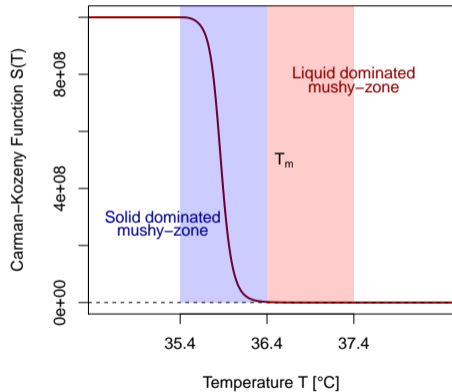
Modified Thermal Conductivity $k(T)$



A - Modeling Functions

Carman-Kozeny Porosity Function $S(T)$

$$S(T) = A_m \frac{(1 - \theta(T))^2}{\theta(T)^3 + \varepsilon} \quad (7)$$

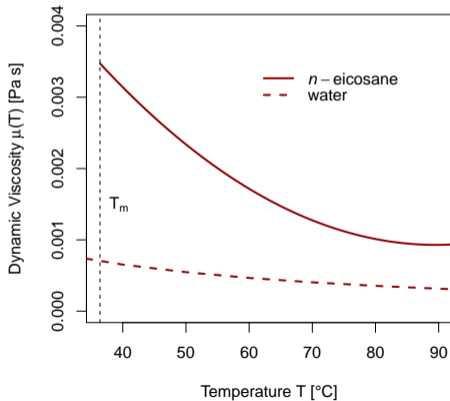


Ali C. Kheirabadi and Dominic Groulx. "Simulating Phase Change Heat Transfer using COMSOL and FLUENT: Effect of the Mushy-Zone Constant". In: *Computational Thermal Sciences: An International Journal* 7.5-6 (2015). DOI: 10.1615/ComputThermalScien.2016014279

$$\mu(T) = (9 \times 10^{-4} T^2 - 0.6529 T + 119.94) \times 10^{-3} \quad (8)$$

A - Modeling Functions

Viscosity of *n*-eicosane $\mu(T)$



A - Modeling Functions

Basic numerical requirements to govern the physics of PCM

conservation equation	solid fraction	liquid fraction
continuity		✓
momentum		✓
energy	✓	✓

→ direct approach: two subdomains for liquid and solid fraction with front tracking algorithm

A - Modeling Functions

Alternative approach: introduction of a mushy zone

- Idea: material properties are smeared out over an user-defined melting temperature range
- Method: use of porosity formulation, liquid and solid co-exist in the mushy zone
- Benefits:
 - avoid numerical singularities
 - use one single mesh
 - easy to implement
- Setback: highly mesh-dependent solution in terms of physical accuracy

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A - Modeling
Functions

Melt Fraction

Gaussian

Heat Capacity

Density

Th. Conductivity

Carman-Kozeny

Viscosity

Requirements

Mushy zone

2D Test-case

BC

Carman-Kozeny

Validation

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E - 2D

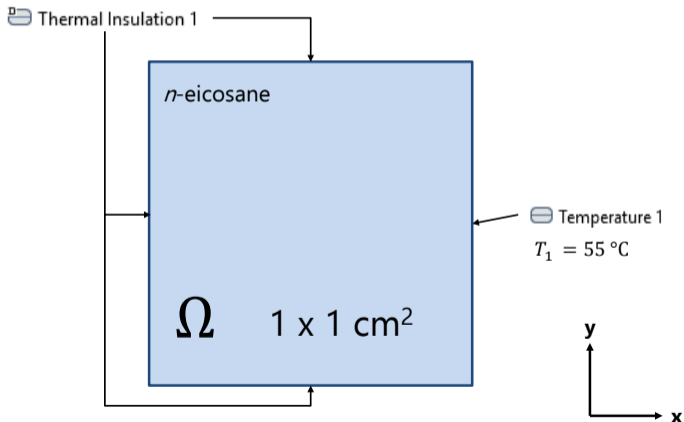
Test-Case

F -

Dimensionless

A - Modeling Functions

2D Test-case



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Carman-Kozeny

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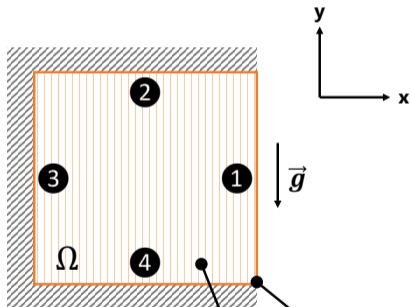
Test-Case

F -

Dimensionless

A - Modeling Functions

Boundary conditions and setup



Boundary	CFD	Heat Transfer
1	No slip wall	$T_R = 40, 55, 70 \text{ } ^\circ\text{C}$
2		Thermal insulation
3		
4		

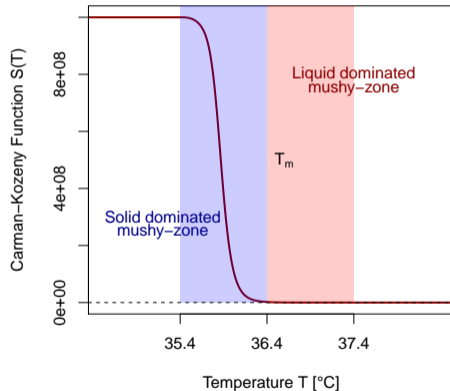
Pressure constraint point, $p = 0 \text{ Pa}$

Transient study, $0 < t < 1000 \text{ s}$

A - Modeling Functions

Carman-Kozeny porosity function $S(T)$

$$S(T) = A_m \frac{(1 - \theta(T))^2}{\theta(T)^3 + \varepsilon} \quad (9)$$



Ali C. Kheirabadi and Dominic Groulx. "Simulating Phase Change Heat Transfer using COMSOL and FLUENT: Effect of the Mushy-Zone Constant". In: *Computational Thermal Sciences: An International Journal* 7.5-6 (2015). DOI: 10.1615/ComputThermalScien.2016014279

B - Comparison to Experimental Data

Results

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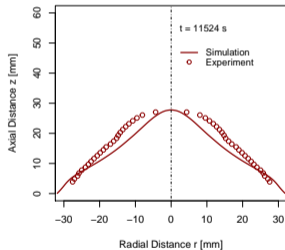
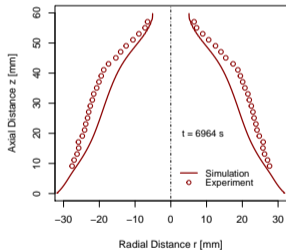
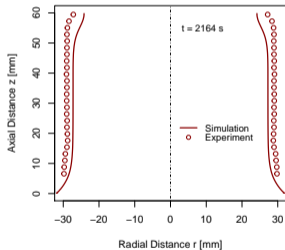
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Material properties

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B - Comparison to Experimental Data

Model Setup

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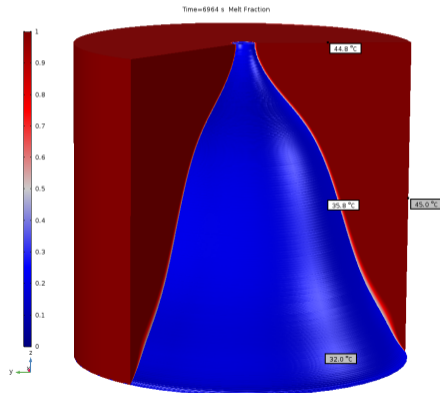
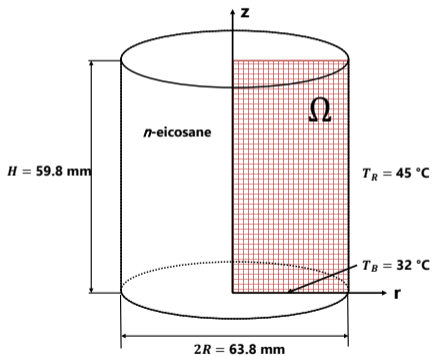
Validation

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B - Comparison to Experimental Data

Material properties of *n*-eicosane, comparison with water

	<i>n</i> -eicosane		water	
	<i>solid</i>	<i>liquid</i>	<i>solid</i>	<i>liquid</i>
density ρ [kg m^{-3}]	910	769	916	997
thermal conductivity k [$\text{W m}^{-1} \text{K}^{-1}$]	0.423	0.146	1.6	0.6
heat capacity C_p [$\text{kJ kg}^{-1} \text{K}^{-1}$]	1.9	2.4	2.1	4.2
melting temperature T_m [$^{\circ}\text{C}$]	36.4	-	0	-
latent heat of fusion L [kJ kg^{-1}]	248	-	334	-

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A - Modeling
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Validation

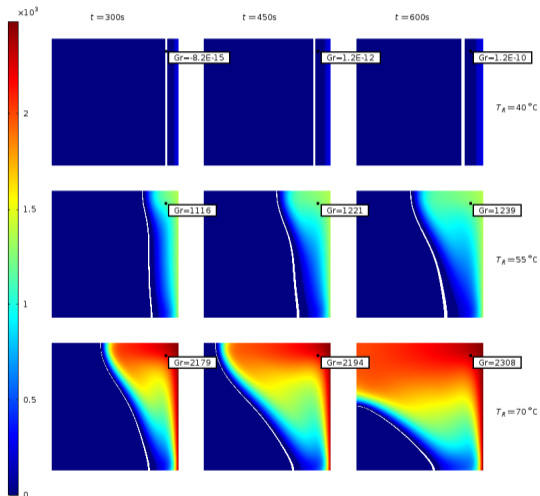
Results

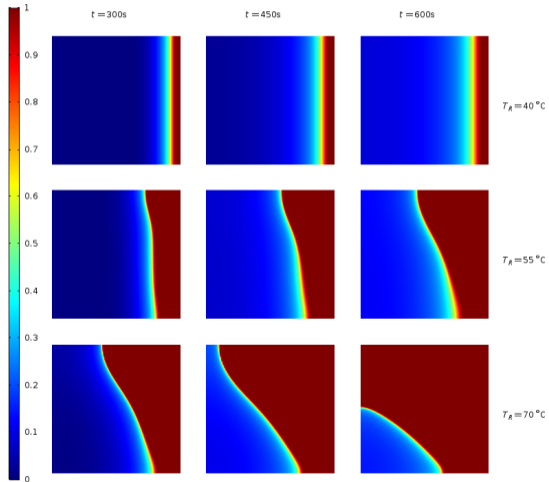
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Material properties

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C - Results
Grashof Number

C - Results
Melt Fraction

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Grashof Number

Melt Fraction

Mesh sensitivity

Melt fraction

Gr

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Validation II

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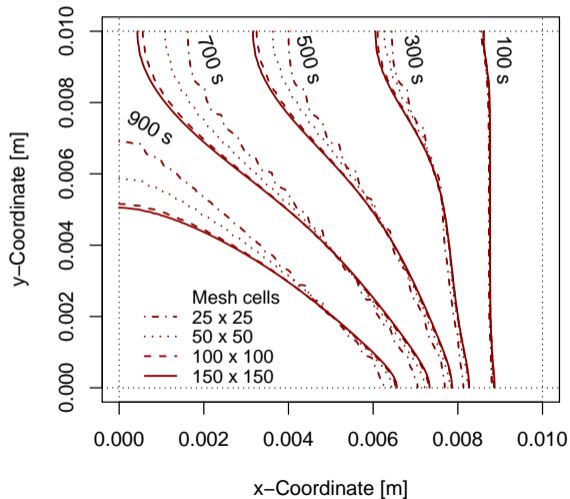
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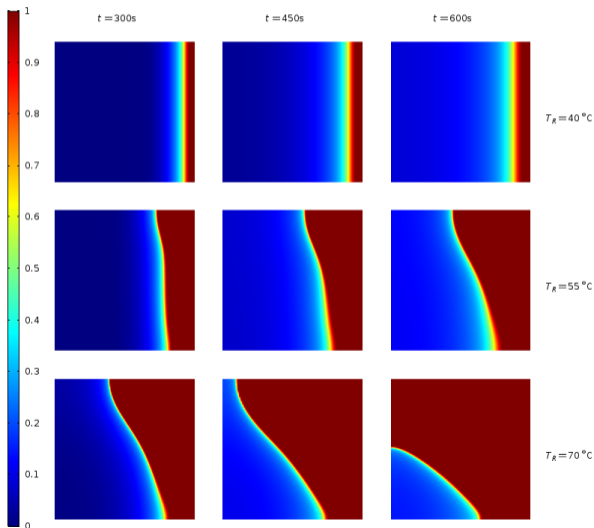
C - Results

Mesh sensitivity - melting front prediction



C - Results

Melt fraction - curvature of melting front



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Melt Fraction

Mesh sensitivity

Melt fraction

Gr

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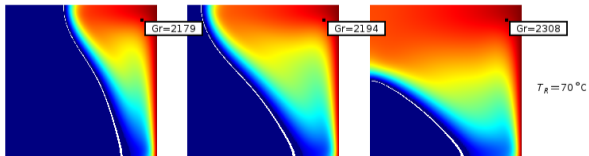
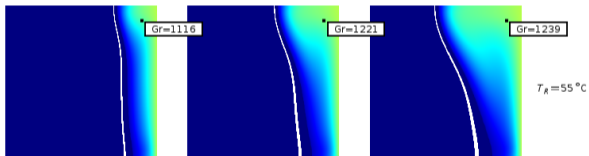
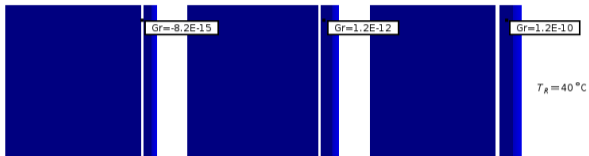
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Local Grashof number - influence of natural convection



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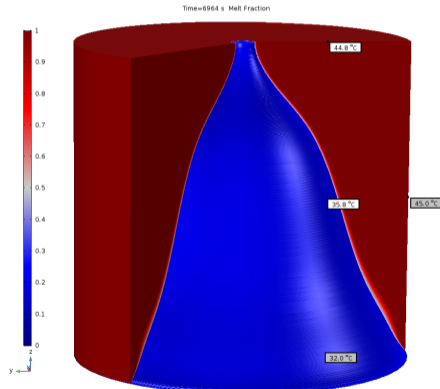
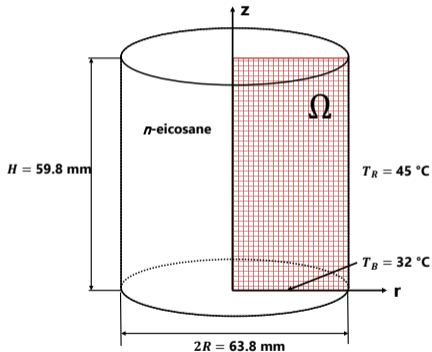
Test-Case

F -

Dimensionless
Estimation

C - Results

Validation case - setup



Benjamin J. Jones et al. "Experimental and numerical study of melting in a cylinder". In: *International Journal of Heat and Mass Transfer* 49.15-16 (2006), pp. 2724–2738

C - Results

Validation case - comparison to experimental data

Appendix

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Gr

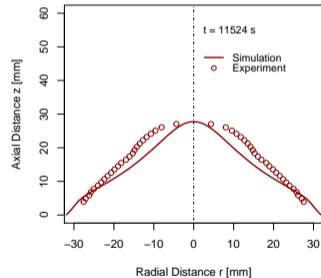
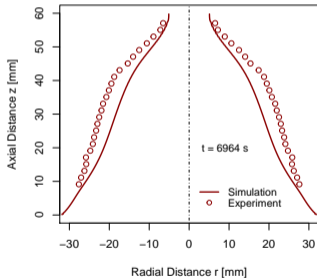
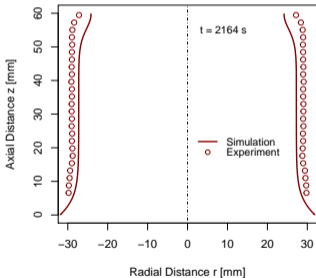
Validation I

Validation II

D -

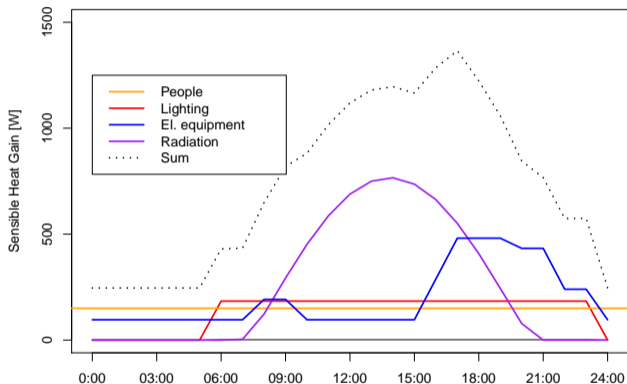
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D - Application Example

Internal Heat Gains, Hourly

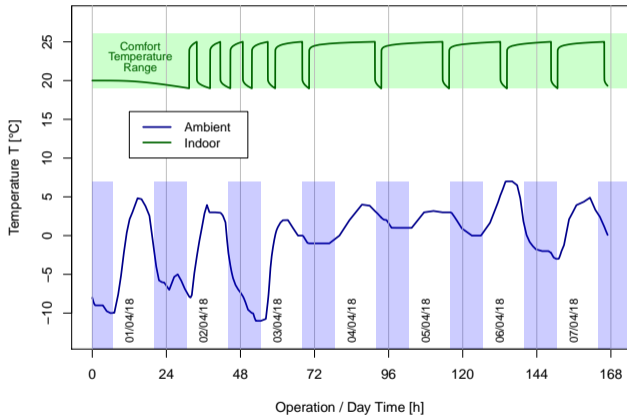


Komit -SN/K-034. *Bygningers energiytelse, Beregning av energibehov of energiforsyning (engl.: Energy performance of buildings, calculation of energy needs and energy supply)*. URL:

<https://www.standard.no/no/Nettbutikk/produktkatalogen/Produktpresentasjon/?ProductID=859500>. 2016

D - Application Example

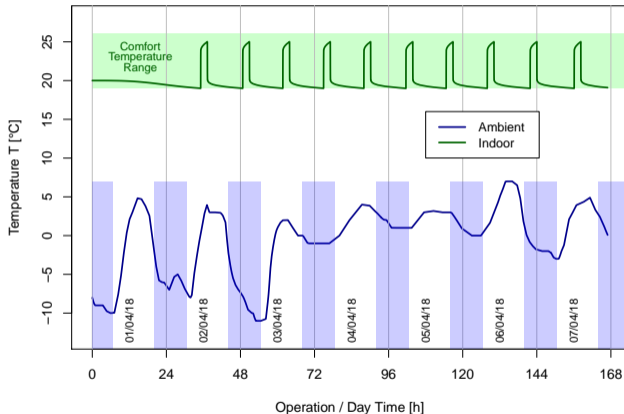
Indoor Temperature w/o PCM



Weather Forecast Oslo. 2018. URL: <https://www.wunderground.com/weather/no/oslo>

D - Application Example

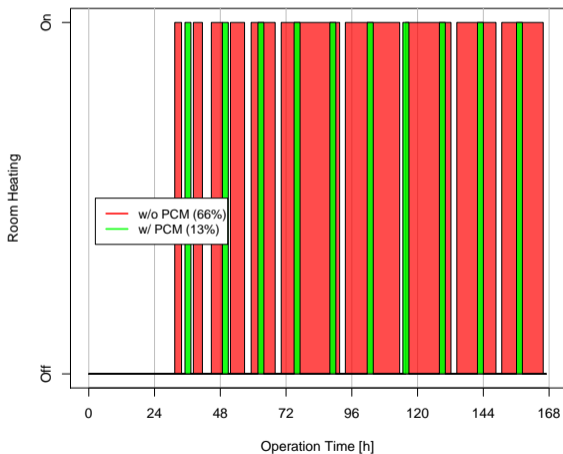
Indoor Temperature w/ PCM



Weather Forecast Oslo. 2018. URL: <https://www.wunderground.com/weather/no/oslo>

D - Application Example

Thermostat



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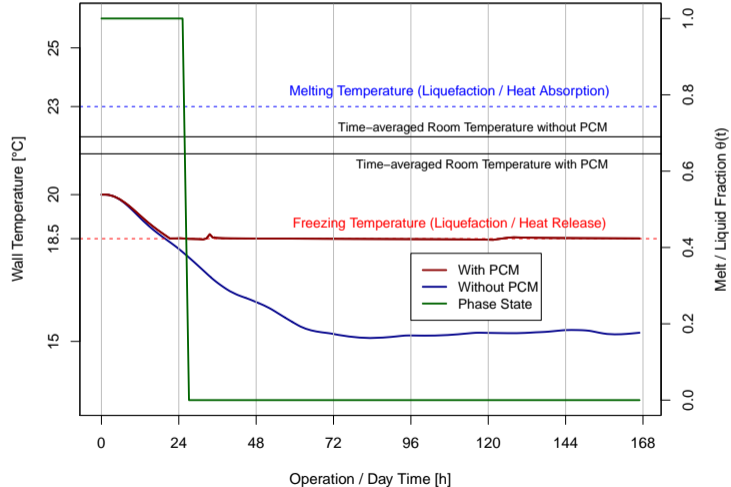
D -
Application
ExampleInternal Heat Gains,
HourlyIndoor Temperature
w/o PCMIndoor Temperature
w/ PCM**Thermostat**Wall Core
Temperature

Wall Cross-Section

E - 2D
Test-CaseF -
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Estimation

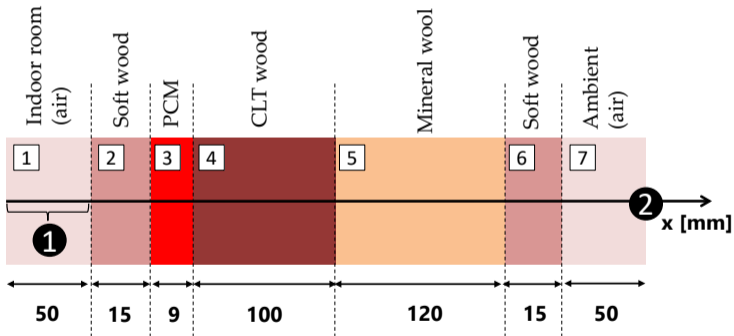
D - Application Example

Wall Core Temperature



D - Application Example

Wall Cross-Section



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ExampleInternal Heat Gains,
HourlyIndoor Temperature
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Test-Case**2D Test-Case**

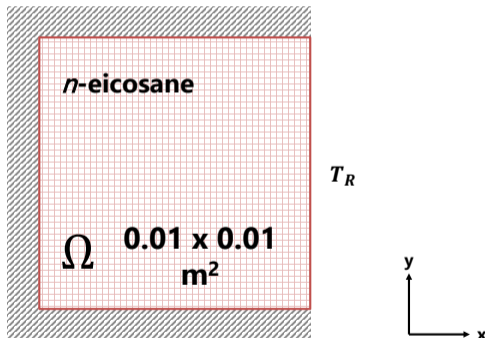
Couplings

2D Test-case

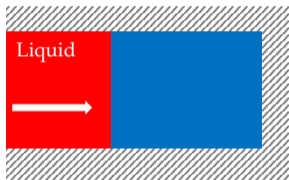
F -
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E - 2D Test-Case

2D Test-Case



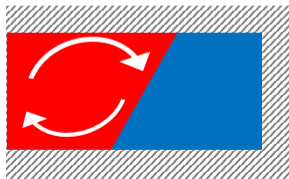
$$T_R > T_m$$



T_R : Boundary Temperature

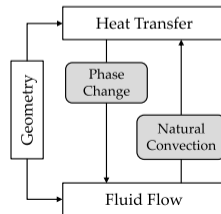
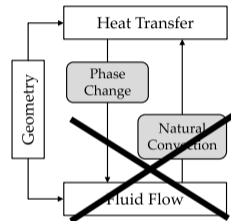
T_m : Melting Temperature

$$T_R \gg T_m$$



E - 2D Test-Case

Multiphysical couplings



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2D Test-Case

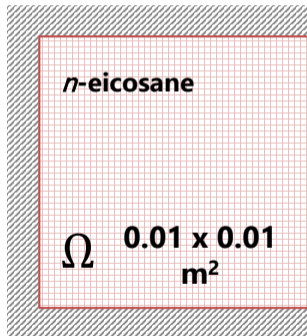
Couplings

2D Test-case

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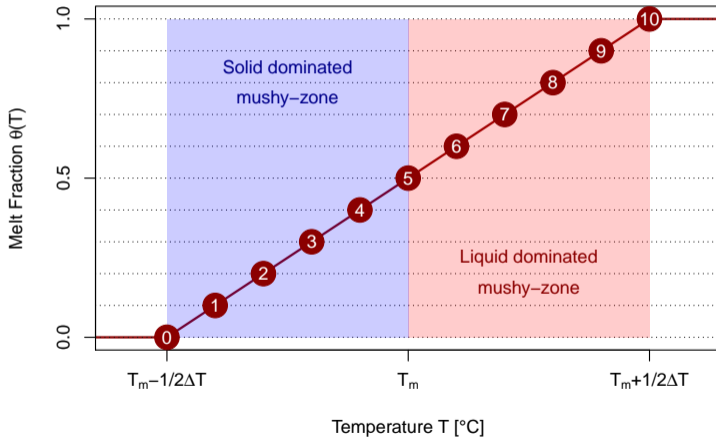
E - 2D Test-Case

2D Test-case

 $T_R = 40, 55, 70 \text{ } ^\circ\text{C}$ 

F - Dimensionless Estimation

Mushy Zone Investigation



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Scaling Variables

Dimensionless

Equations

Dimensionless
NumbersValues For Liquid
Fraction

F - Dimensionless Estimation

Scaling Variables

$$\tilde{x} = \frac{x}{H}$$

$$\tilde{y} = \frac{y}{H}$$

$$\tilde{p} = \frac{p - p_{ref}}{\rho u_0^2}$$

$$\tilde{t} = \frac{u_0 t}{H}$$

$$\tilde{\mathbf{u}} = \frac{\mathbf{u}}{u_0}$$

$$\tilde{T} = \frac{T - T_{ref}}{T_R - T_{ref}}$$

$$\tilde{\Phi}_v = \left(\frac{H}{u_0}\right)^2 \Phi_v$$

$$\tilde{\nabla} = H \nabla$$

$$\frac{D}{D\tilde{t}} = \left(\frac{H}{u_0}\right) \frac{D}{Dt}$$

F - Dimensionless Estimation

Dimensionless Equations

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Test-CaseF -
Dimensionless
EstimationMushy Zone
Investigation

Scaling Variables

Dimensionless
EquationsDimensionless
NumbersValues For Liquid
Fraction

$$\tilde{\nabla} \cdot \tilde{\mathbf{u}} = 0 \quad (10)$$

$$\frac{D\tilde{\mathbf{u}}}{D\tilde{t}} = -\tilde{\nabla}\tilde{p} + \left[\frac{\mu}{u_0\rho H} \right] \tilde{\nabla}^2 \tilde{\mathbf{u}} - \left[\frac{g\beta(T_R - T_m)H}{u_0^2} \right] \left(\frac{\mathbf{g}}{g} \right) (\tilde{T} - \tilde{T}_m) \quad (11)$$

$$\frac{D\tilde{T}}{D\tilde{t}} = \left[\frac{k}{u_0\rho HC_p} \right] \tilde{\nabla}^2 \tilde{T} + \left[\frac{\mu u_0}{\rho HC_p(T_R - T_m)} \right] \tilde{\Phi}_v \quad (12)$$

F - Dimensionless Estimation

Dimensionless Numbers

Brinkman $Br = \frac{\mu u_0^2}{k\Delta T}$

heat production visc. dissipation vs heat transport by cond.

Grashof $Gr = \frac{g\beta\Delta TH^3}{\nu^2}$

buoyant forces vs viscous forces

Prandtl $Pr = \frac{\nu}{\alpha}$

momentum diffusivity vs thermal diffusivity

Rayleigh $Ra = \frac{g\beta\Delta TH^3}{\alpha\nu} = GrPr$

heat transport conv. vs cond.

Reynolds $Re = \frac{\rho u_0 H}{\mu}$

inertial vs viscous forces

F - Dimensionless Estimation

Values For Liquid Fraction

Dimensionless group	T_R		
	40 °C	55 °C	70 °C
$\left[\frac{\mu}{u_0 \rho H} \right] = \frac{1}{\text{Re}}$	1	1	1
$\left[\frac{g \beta (T_R - T_m) H}{u_0^2} \right] = \frac{\text{Gr}}{\text{Re}^2} = \frac{\text{Ra}}{\text{PrRe}^2}$	266	1376	2486
$\left[\frac{k}{u_0 \rho H C_p} \right] = \frac{1}{\text{RePr}}$	0.008	0.008	0.008
$\left[\frac{\mu u_0}{\rho H C_p (T_R - T_m)} \right] = \frac{\text{Br}}{\text{RePr}}$	1.25×10^{-10}	2.42×10^{-11}	1.34×10^{-11}

Appendix

A - Modeling
Functions

Validation

C - Results

D -
Application
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