

Comsol Multiphysics as a General Platform for the Simulation of Complex Electrochemical Systems

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Overview

- Basics of electrochemical simulation and modeling
- State of the art
- Needs for a general tool
- Application of Comsol Multiphysics

Basics of electrochemical simulation

Electrochemistry is really multiphysics!

Diffusion

Diffusion convection

Migration

More Exotic:

Magnetohydrodynamics

Maxwell's electrochemistry

Phase-field modelling

Closed form solutions of transport equation don't exist for most of these problems

Need for discretization for the non-linear nature of the problems
geometry and boundary conditions



State of the art

Finite Differences

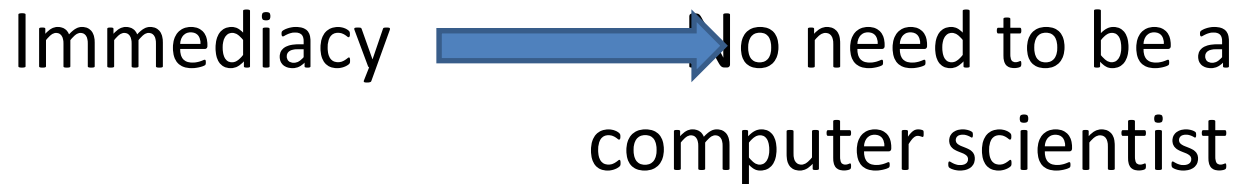
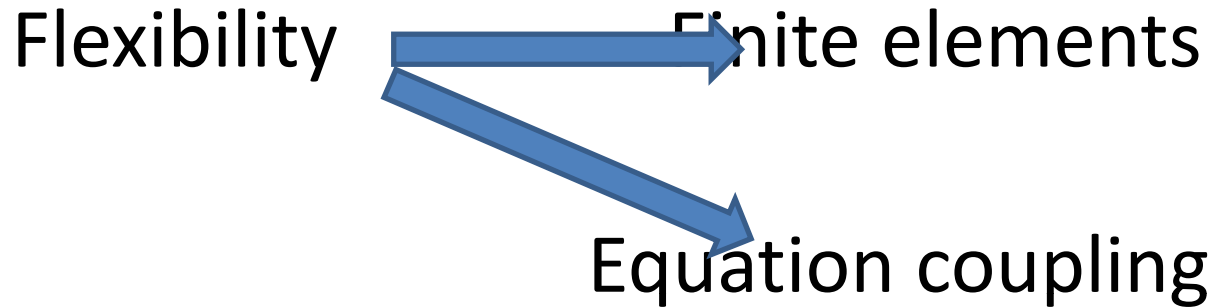
- Easy to implement but difficult to extend to multidimensional problems for general boundary conditions

Finite Elements

- Difficult to implement respect the FD
- More general as they allow the straightforward implementation for the simulation of complex shapes in 2D and 3D.
- Theoretical framework is very well developed allowing error estimation and adaptive techniques

The use of finite elements in electrochemistry shows more generality and it's the best choice for a tool aiming at general electrochemical simulations.

Needs for a general tool



Applications of COMSOL multiphysics/1

Microelectrode arrays simulation

- 1) Cyclic Voltammetry (CV) at microelectrodes array
- 2) Electrodes in the diffusion convection regimes
- 3) Magnetochemistry
- 4) Phase Field modelling of electrodeposition

Applications of COMSOL Multiphysics/2

Cyclic Voltammetry (CV) at microelectrodes array

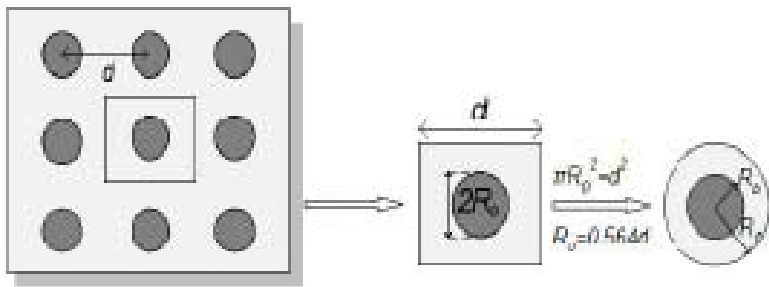


Figure 1. Microelectrode array and the domain wall approximation.

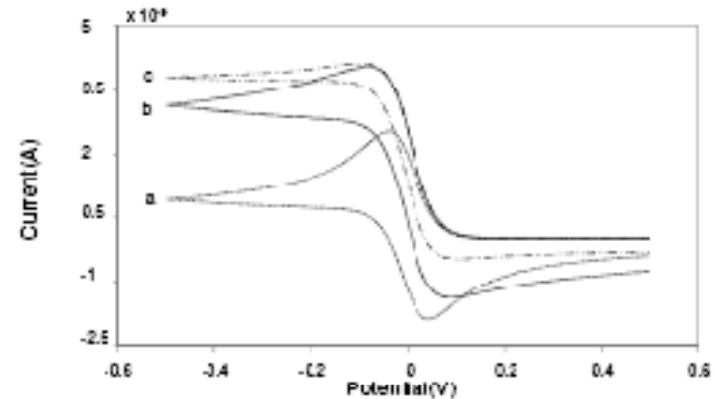
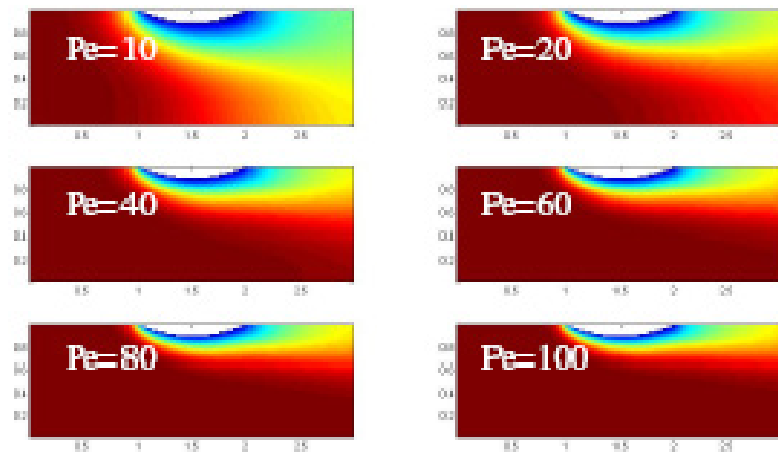


Figure 2. CV's simulated on a microelectrode array with (a) $R_b = 10 \mu\text{m}$ and $R_o = 20 \mu\text{m}$, (b) $R_b = 10 \mu\text{m}$ and $R_o = 50 \mu\text{m}$ and (c) $R_b = 10 \mu\text{m}$ and $R_o = 100 \mu\text{m}$

Applications of COMSOL Multiphysics/3

Electrodes in the diffusion convection regimes



| Peclet Number | Coll. Eff | SS Time Inlaid | Coll Eff | SS Time Bumped |
|---------------|-----------|----------------|----------|----------------|
| 10 | 41.2 | 3.1 | 43.8 | 3.0 |
| 20 | 26.4 | 2.4 | 28.1 | 2.3 |
| 40 | 16.6 | 2.3 | 17.7 | 2.3 |
| 60 | 12.6 | 2.4 | 13.5 | 2.3 |
| 80 | 10.5 | 2.5 | 11.1 | 2.4 |
| 100 | 9.00 | 2.7 | 9.55 | 2.5 |

Figure 3. Concentration distribution in the stationary diffusion convection regimes for the bumped electrode.

Table 1: collection efficiency and current stabilization times dependence on the Peclet number for bumped and inlaid microelectrodes.

Applications of COMSOL multiphysics/4

Magnetochemistry

Reverse Flow
Occurrence at High
Magnetohydrodynamic
Numbers

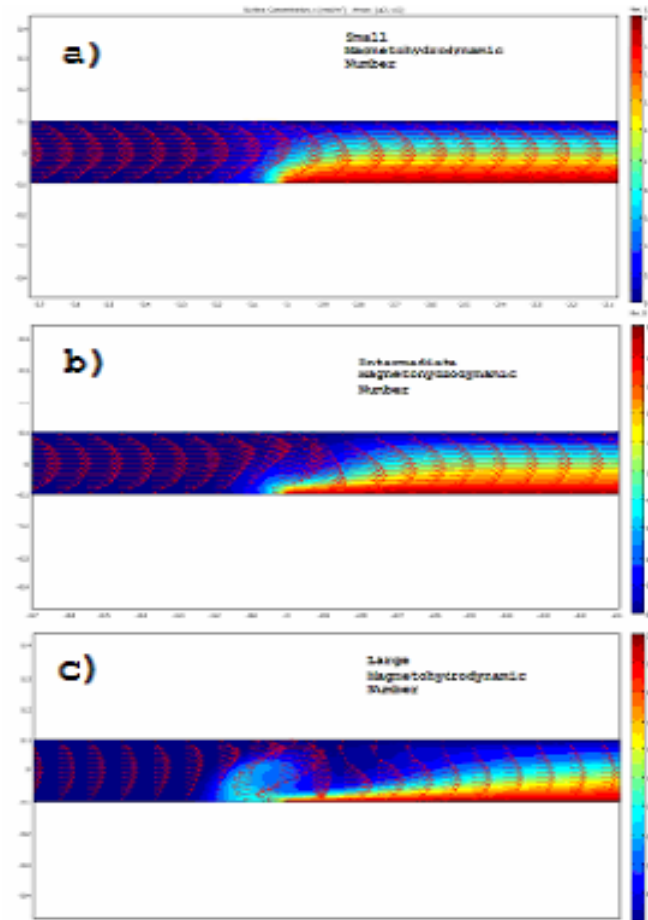
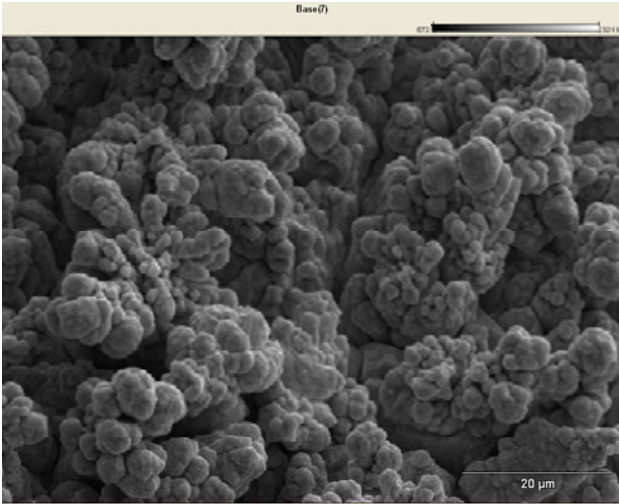


Figure 4. Magnetohydrodynamic convection and reverse flow occurrence in a parallel plates

Applications of COMSOL multiphysics/4

Phase-field modeling of electrodeposition



Electrodeposited Copper

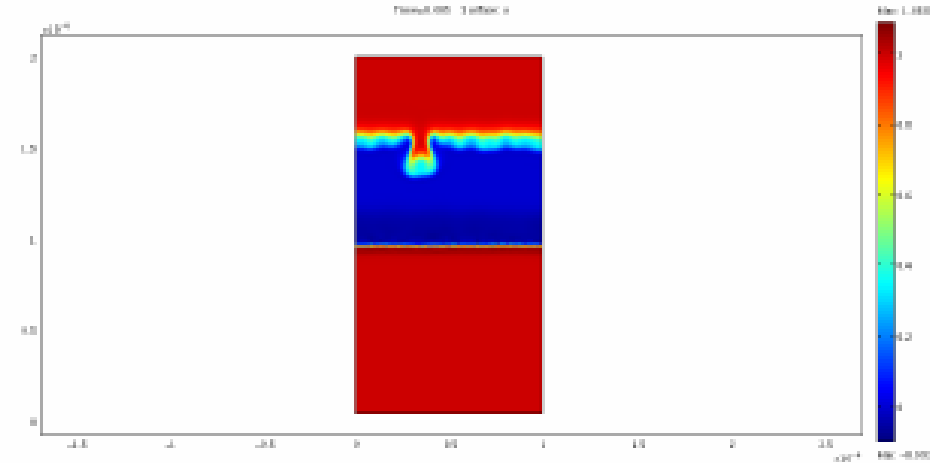


Figure 5. Break-up of morphological stability in the electrodeposition. First stage of the dendritic growth.

Conclusions

- COMSOL multiphysics has been successfully applied to the simulation of a variety of electrochemical experiments
- The same environment provide the possibility of simulating experiments in the whole range of electrochemistry